

REMARKS

Reconsideration of this patent application is respectfully requested in view of the following remarks.

The Applicants comment upon the prior art rejections of the claims as follows:

On Page 2 of the Office Action, the Patent Examiner has rejected claims 1-9 and 11 under 35 U.S.C. 103(a) as being unpatentable over *Pirooz et al* (EP 0701275) in view of *Verhaverbeke et al* (U.S. Patent No. 6,132,522).

On Page 6 of the Office Action, the Patent Examiner has rejected claims 12-15 under 35 U.S.C. 103(a) as being unpatentable over *Pirooz* (EP 0701275) in view of *Verhaverbeke* (U.S. Patent No. 6,132,522) as applied to claims 1-9 and 11, and further in view of *Berman* (U.S. Patent No. 5,014,737) (newly cited), or *Davison et al* (U.S. Patent No. 5,593,538) (newly cited).

The present invention is directed to a process for the wet chemical treatment of semiconductor wafers with treatment liquids in baths, consisting of the steps of

firstly treating the semiconductor wafers in a bath with an aqueous HF solution containing HF and optionally HCl and optionally a surfactant;

then treating the semiconductor wafers in a bath with an aqueous O₃ solution containing O₃ and optionally HF; and

then treating the semiconductor wafers in a bath with an aqueous HCl solution containing HCl and optionally O₃;

whereby these treatment steps form a treatment sequence B₂, which avoids rinsing with water or another treatment liquid and the addition of fresh water or other liquids to the treatment baths.

Also, the present invention is directed to a process for the wet chemical treatment of semiconductor wafers with treatment liquids in baths, comprising the steps of

firstly treating the semiconductor wafers in a bath with an aqueous HF solution containing HF and optionally HCl and optionally a surfactant;

then treating the semiconductor wafers in a bath with an aqueous O₃ solution containing O₃ and optionally HF; and

then treating the semiconductor wafers in a bath with an aqueous HCl solution containing HCl and optionally O₃;

whereby these treatment steps form a treatment sequence B₂; and

circulating the treatment liquids of said baths by taking a part from each of said baths, filtering and returning the part to the corresponding treatment bath.

Furthermore, the present invention is directed to a process for the wet chemical treatment of semiconductor wafers with

treatment liquids in baths, comprising the steps of
firstly treating the semiconductor wafers in a bath with an
aqueous HF solution containing HF and optionally HCl and
optionally a surfactant;

then treating the semiconductor wafers in a bath with an
aqueous O₃ solution containing O₃, and optionally HF; and

then treating the semiconductor wafers in a bath with an
aqueous HCl solution containing HCl and optionally O₃; and

circulating the treatment liquids of said baths by taking a
part from each of said baths, filtering and returning the part to
the corresponding treatment bath;

whereby these treatment steps form a treatment sequence B₂,
which avoids rinsing with water or another treatment liquid and
the addition of fresh water or other liquids to the treatment
baths.

In response to these rejections, it appears that there is
still a misunderstanding of the present invention which is based
upon a misinterpretation of the teachings of the prior art
references. Specifically, the primary reference *Pirooz* only
teaches a chemical treatment process in which there are two (2)
active, positive steps for metal removal. This begins with the
last paragraph in column 2 in *Pirooz* in which an aqueous HF
treatment solution is used. Then *Pirooz* includes column 3, lines

8-13, wherein a first water rinsing step is disclosed. This is followed by the disclosure in *Pirooz* column 3, lines 25-35, in which an aqueous ozone treatment step is disclosed. Then there is a second water rinsing step in *Pirooz* column 3, lines 35-40.

The *Verhaverbeke* secondary reference discloses the elimination of water rinse (column 3, lines 15 to 23) between treatments with various chemical reagents for semiconductor cleaning. It is respectfully submitted that there is no explanation given in the Office Action that satisfactorily explains why one skilled in the art would modify the *Pirooz* primary reference to add a third treatment step with a third active chemical reagent, other than through the use of hindsight based upon the Applicants' own teachings and the Applicants' own disclosure in the present Specification.

On page 7 of the Office Action, the Patent Examiner recognizes that obviousness can only be established by combining or modifying the teachings of the prior art to produce the claimed invention where there is some teaching, suggestion, or motivation to do so found either in the references themselves or in the knowledge generally available to one of ordinary skill in the art. Further, the Patent Examiner points out that regarding the present application, *Verhaverbeke* teaches the elimination of a DI rinse between each chemical treatment step results in significant savings and higher output.

Finally, the Patent Examiner concludes that the higher output and increased saving is sufficient motivation to a person of ordinary skill in the art at the time of the invention to modify *Pirooz* by eliminating the DI rinse, as taught by Verhaverbeke.

If this argumentation is accepted, *Pirooz* would teach a process which comprises firstly treating semiconductor wafers with an aqueous HF solution containing HF; and then treating the semiconductor wafers in a bath with an aqueous O₃ solution containing O₃.

Pirooz would not teach treating the semiconductor wafers in a bath with an aqueous HCl solution containing HCl. This is also accepted by the Patent Examiner on Page 3 of the Office Action. However, at this stage, the argumentation of the Patent Examiner becomes more non-convincing. The Patent Examiner argues that Verhaverbeke teaches that a rinse fluid may be DI water or a very dilute aqueous solution of hydrochloric acid to prevent metallic deposition on the surface (see page 8 of the Office Action). The Patent Examiner then concludes that this teaching would be sufficient motivation to modify *Pirooz* by using a solution of hydrochloric acid as a rinse in place of the DI rinse to prevent metallic deposition.

This is an apparent wrong conclusion, since Verhaverbeke teaches a DI rinse and a rinse with a very dilute aqueous

solution of hydrochloric acid as equivalent steps (col. 5, lines 5-9). *Verhaverbeke* does not contain any teaching or motivation for preferring a rinse with a very dilute aqueous solution of hydrochloric acid to a rinse with DI. Therefore, if the above argument is accepted that *Verhaverbeke* suggests eliminating a DI rinse, the same argument must hold for the rinse with a very dilute aqueous solution of hydrochloric acid. Thus, *Verhaverbeke* clearly does not suggest modifying *Pirooz* by using a solution of hydrochloric acid, because this step would act in opposition to a higher output and an increased savings.

In addition to all of the above, it is respectfully submitted that any attempt to combine the teachings of the *Pirooz* primary reference and the *Verhaverbeke* secondary reference would destroy the teachings of the *Pirooz* primary reference. This is impermissible under 35 U.S.C. 103.

Pirooz in column 1, in lines 55 to 59 and in column 2, in lines 1 to 2, discloses that the process of *Pirooz* comprises the steps of contacting the surface of the silicon wafer with an aqueous solution containing hydrofluoric acid to remove metals from the wafer surface, and contacting the hydrofluoric acid treated wafers with ozonated water to grow a hydrophilic oxide layer on the surface of the silicon wafer.

Pirooz in column 3, in lines 14 to 30, discloses that the rinsed wafers are contacted with high purity ozonated water,

i.e., deionized water containing ozone and having a concentration of no more than about 0.01 parts per billion (weight/volume) of each of iron, chromium, titanium and other metals which will degrade minority carrier lifetime, to produce a wafer having a hydrophilic oxide surface. The resulting silicon oxide layer preferably has a thickness of about 0.6 to about 1.5 nanometers and a concentration of no more than 1×10^9 , more preferably no more than about 1×10^8 atoms/cm² of each of iron, chromium, titanium and other metals which degrade minority carrier lifetime. Silicon oxide layers of this thickness and purity may be produced by immersing the wafers in a bath containing about 0.1 parts per million (ppm) to 50 ppm of ozone, preferably about 10 to about 15 ppm ozone, at a temperature of about 0°C to about 60°C, preferably room temperature, for a period of at least about 0.1 minutes, preferably about 2 to about 10 minutes.

Therefore, any further treatment of the hydrophilic oxide layer on the surface of the silicon wafer of *Pirooz* as taught by *Verhaverbeke* would destroy the properties of the surface, which surface is required by *Pirooz*. Such a radical reconstruction of the prior art references that will destroy the hydrophilic oxide layer surface as mandated by *Pirooz* is impermissible according to 35 U.S.C. 103. This destruction of the teachings of the primary reference to *Pirooz* prevents the Patent Examiner from establishing a *prima facie* case of obviousness under 35 U.S.C. 103.

Therefore, the Patent Examiner must provide an explanation as to why a person skilled in the art would have modified *Pirooz*, based upon *Verhaverbeke* to produce the claimed three step process devoid of any water rinsing steps.

The comparative testing on pages 4 and 5 of the Specification shows a significant improvement in reducing LPD (lightpoint defects).

The comparison below of an Invention example with comparative examples shows the advantageous effect of the present invention.

The Table contains the results (3 sigma values) of conventional counts of particles of specific size (LPD=light point defects), which were in each case carried out on 20 polished semiconductor wafers after a wet chemical treatment sequence and drying which was the same for all the semiconductor wafers. The attached Table appears on page 5 of the present Specification.

The example (B) of the Invention comprised the treatment sequence: HF bath, deionized water/ozone bath, HCl bath with megasonic exposure.

Comparative Example 1 (C1) comprised the treatment sequence: HF bath, deionized water/ozone bath, rinsing with deionized water outside the bath. This corresponds to *Pirooz*, set "B" without megasonics.

Comparative Example 2 (C2) comprised the treatment sequence:

HF bath, deionized water/ozone bath with subsequent rinsing using deionized water in the bath and megasonic exposure. This corresponds to *Pirooz* set "A" with megasonics.

Berman (U.S. Patent No. 5,014,737) in column 2, lines 50 to 68, discloses a recirculating chemical bath arrangement which is designed to overcome the deficiencies with regard to the two basic forms of quartz tank recirculating systems. This includes a quartz process tank having a trough extending about the top portion of the outer walls of the tank. The system also includes a pump/filter circuit which pumps the chemical solution into the bottom of the process tank, the liquid flowing over the tank walls into the trough. Also the process tank is provided with heater units secured to outer surface thereof. However, rather than connecting the pump/filter input directly to the trough, there is a second, smaller chamber secured to the end of the process tank and disposed to receive liquid flow from the trough as well as from the process tank. The second chamber includes a fitting at the bottom thereof to supply liquid to the pump/filter circuit.

This process and apparatus of *Pirooz* in view of *Berman* will produce massive contamination of the treatment baths, leading to the high increase in LPD's for C1 and C2, as shown in the attached Table. For example, according to the Invention, the LPD's are 7, or 30, depending upon the size of the LPD. For *Pirooz* in view of *Berman*, the corresponding number of LPD's would

be 13 for C1 and would be 50 for C2, if B of the Invention were 7. Again, for *Pirooz* in view of *Berman*, the corresponding number of LPD's would be 50 for C1 and would be 140 for C2, if B were 30. Thus, one skilled in the art would never combine the teachings of *Berman* with either *Pirooz* alone, or *Pirooz* in combination with *Verhaverbeke*. To do so would produce the extremely undesirable results discussed above, in which the greatly increased number of LPD's produced is completely undesirable.

Davison (U.S. Patent No. 5,593,538) in column 1, lines 54 to 67 and column 2, in lines 1 to 5, discloses a method for etching a substrate. More particularly, *Davison* provides a wet etching process for an oxide layer on a semiconductor material.

In a wet etching process in accordance with *Davison*, polycrystalline silicon is etched when it is exposed to an etchant. The etching of the polycrystalline silicon may cause open polycrystalline silicon lines in some areas as well as reliability problems by creating very thin polycrystalline silicon lines. An objective of *Davison* is to provide an etching process that does not etch a conductive structure such as, for example, a polycrystalline silicon structure, on a semiconductor wafer. Through an extensive investigation, it has been discovered that the etching of the polycrystalline silicon structure is caused by oxygen dissolved in the etchant.

Therefore, *Davison* provides a method for protecting the conductive structure on a substrate during a wet etching process by purging away oxygen dissolved in the etchant.

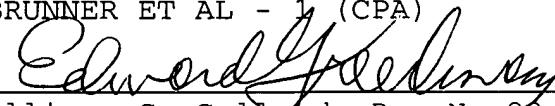
This process and apparatus of *Pirooz* in view of *Davison* will produce massive contamination of the treatment baths leading to the huge increase in LPD's for C1 and C2, as shown in the attached Table. For example, according to the Invention, the LPD's are 7, or 30, depending upon the size of the LPD. For *Pirooz* in view of *Davison*, the corresponding number of LPD's would be 13 for C1 and would be 50 for C2, if B of the Invention were 7. Again, for *Pirooz* in view of *Davison*, the corresponding number of LPD's would be 50 for C1 and would be 140 for C2, if B were 30. Thus, one skilled in the art would never combine the teachings of *Davison* with either *Pirooz* alone, or *Pirooz* in combination with *Verhaverbeke*. To do so would produce the extremely undesirable results discussed above, in which the greatly increased number of LPD's produced is completely undesirable.

In conclusion, the present invention is believed to be patentable over all the prior art references applied by the Patent Examiner under 35 U.S.C. 103. A prompt notification of

allowability is respectfully requested.

Respectfully submitted,

BRUNNER ET AL - 1 (CPA)

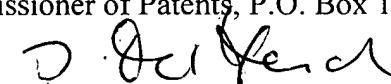

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Enclosure: Copy of Table

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